

## **Course: Organic Chemistry and Soil Coordination Compounds**

### **Identification**

**Code:** SOL 859

**Credits:** 4 (3 hours theory -1 hour practice)

**Level:** Master and Doctorate

**Professor:** Leandro Souza da Silva

**System:** Annual (II semester)

### **Discipline objectives**

To provide students with knowledge of the basic concepts of organic chemistry to relate them to the nature of organic matter and its biological, physical, and chemical properties. To provide students with knowledge of coordination compounds in soil chemistry to identify, characterize, and relate the factors that act on the formation, structuring, and stability of organic and inorganic complexes in the soil solution and at the solid-solution interface. To provide a theoretical foundation to understand soil and water quality, agricultural production, and agroecosystem sustainability.

### **Syllabus**

Fundamentals of coordination compounds and organic chemistry; coordination compounds in soil solution; coordination compounds at the solid-solution interface; chemistry of soil organic matter; evaluation and characterization of organic substances; interactions of organic matter in the soil.

### **Methodology and/or teaching instruments**

Lectures, individual seminars, discussion of scientific articles (overhead projector and blackboard).

### **Forms of evaluation**

Written exams, seminars, activity reports, and class participation.

### **Program - Title and Breakdown of Units**

#### **Unit 1**

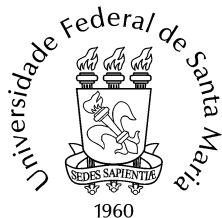
##### **Fundamentals of organic chemistry**

- 1.1 - Introduction to organic chemistry
  - 1.1.1 - Hybridization of carbon orbitals
  - 1.1.2 - Structural and geometric representations
- 1.2 - Nomenclature and characterization
  - 1.2.1 - Alkanes
  - 1.2.2 - Alkenes
  - 1.2.3 - Alkynes
  - 1.2.4 - Aromatic compounds
- 1.3 - Main groups of organic compounds
  - 1.3.1 - Halides
  - 1.3.2 - Alcohols
  - 1.3.3 - Phenols
  - 1.3.4 - Ethers
  - 1.3.5 - Amines
  - 1.3.6 - Aldehydes and ketones
  - 1.3.7 - Carboxylic acids

#### **Unit 2**

##### **Chemistry of soil organic matter**

- 2.1 - Importance and functions of organic matter
- 2.2 - Formation and decomposition of organic matter



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2.2.1 - Organic matter compartments and their characterization

2.2.2 - Physical and chemical fractionation of organic matter and its relationship with soil dynamics

2.3 - Structure and reactions of organic matter and its relationship with soil properties

### **Unit 3**

#### **Evaluation and characterization of organic substances**

3.1 - Quantitative analysis of carbon in soil and solution

3.2 - Using spectroscopic techniques to characterize organic matter

3.2.1 - UV-Visible

3.2.2 - Infrared

3.2.3 - NMR and EPR

### **Unit 4**

#### **Fundamentals of coordination compounds**

4.1 - Introduction to coordination compounds

4.1.1 - Importance of coordination compounds and their relationship with the environment

4.1.2 - Historical aspects of coordination compounds

4.2 - Basic concepts

4.2.1 - Complex, central atom, and ligands

4.2.2 - Types of ligands and coordination numbers

4.3 - Models in the study of coordination compounds

4.3.1 - Valence bond theory

4.3.2 - Crystal field theory

4.3.3 - Molecular orbital theory

### **Unit 5**

#### **Coordination compounds in soil solution**

5.1 - Cations in aqueous media and their dynamics in the soil solution

5.1.1 - Coordinated cations with water

5.1.2 - Oxo-hydroxo and aquo-ion forms

5.2 - Types of complexes in the soil solution and interactions

5.2.1 - "Free" ions

5.2.2 - Outer-sphere complexes (ionic pairs)

5.2.3 - Inner-sphere complexes

5.3 - Main inorganic complexes in solution

5.4 - Main organic complexes in solution

5.5 - Speciation in the soil solution

5.5.1 - Importance of speciation in the activity of nutrients and toxic elements

5.5.2 - Soil solution extraction and characterization

5.5.3 - Use of computer programs for soil solution speciation

### **Unit 6**

#### **Coordination compounds at the solid-solution interface**

6.1 - Characteristics of the surface of soil particles

6.1.1 - Charges and functional groups

6.1.2 - Protonation

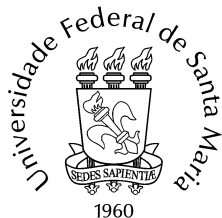
6.2 - Surface complexes and the availability of nutrients and toxic elements

6.2.1 - Outer-sphere complexes (exchange reactions)

6.2.2 - Inner-sphere complexes (specific adsorption)

6.3 - Coordination compounds and the solubility of minerals in the soil

6.3.1 - Dissolution by protonation



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### 6.3.2 - Dissolution by binders

#### Unit 7

##### **Organic matter interactions in soil**

7.1 - Interactions of organic matter with inorganic colloids and their relation with soil dynamics

7.2 - Interactions of organic matter with anthropogenic compounds and their relationship with the environment

#### Recommended literature

ALLINGER, N.L.; CAVA, M.P.; DE JONGH, D.; JOHNSON, C.R. ; LEBEL, N.A.; STEVENS, C.A. **Química Orgânica**. Rio de Janeiro: Editora Guanabara Koogan S.A. 1978. 322p.

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SANTOS, G.A.; SILVA, L.S.; CANELLAS, L.P.; CAMARGO, F.A.O, (Eds.) **Fundamentos da matéria orgânica do solo: ecossistemas tropicais e subtropicais**, 2 ed. Porto Alegre: Metrópole, 2008. 636p.

COLEMAN, D.C., OADES, J.M., UEHARA, G. **Dynamics of soil organic matter in tropical ecosystems**. Honolulu: University of Hawaii, 1989. 524 p.

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HUANG, P.M.; SCHNITZER, M. (Eds). **Interactions of soil minerals with natural organics and microbes**. Madison: Soil Science Society of America, 1986. 606p.

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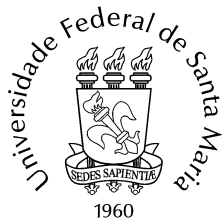
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SILVERSTEIN, R.M., BASSLER, G.C., MORRILL, T.C. **Identificação espectrométrica de compostos orgânicos**. Rio de Janeiro: Editora Guanabara, 1979. 299 p.

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STEVENSON, F.J. **Humus chemistry: genesis, composition, reaction**. (2<sup>nd</sup> ed.) New York: J. Wiley, 1994. 496p.

STUMM, W; MORGAN, J.J. **Aquatic chemistry**. (3<sup>th</sup> ed.) New York: John Wiley & Sons. 1996. 1022p.

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